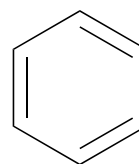
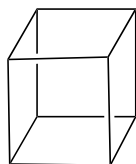




Validating Eaton's Hypothesis: Cubane as a Benzene Bioisostere

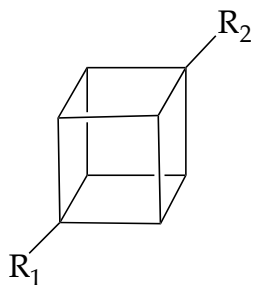
Chalmers, B. A.; Xing, H.; Houston, S.; Clark, C.; Ghassabian, S.; Kuo, A.; Cao, B.; Reitsma, A.; Murray, C. E.; Stok, J. E.; Boyle, G. M.; Pierce, C. J.; Littler, S. W.; Winkler, D. A.; Bernhardt, P. V.; Pasay, C.; De Voss, J. J.; McCarthy, J.; Parsons, P. G.; Walter, G. H.; Smith, M. T.; Cooper, H. M.; Nilsson, S. K.; Tsanaktsidis, J.; Savage, G. P.; Williams, C. M., Validating Eaton's Hypothesis: Cubane as a Benzene Bioisostere. *Angew Chem Int Ed Engl* **2016**, 55 (11), 3580-5.



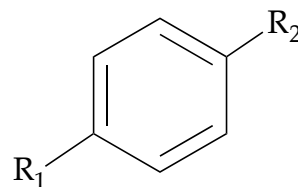
Michael Houghton
Wipf Group
07/09/16

Bioisosteres and Eaton's Hypothesis

Eaton's Hypothesis: Cubane may be a suitable bioisostere of Phenyl due to it having a similar size and shape

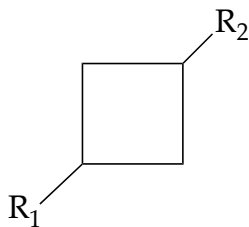


$$C_{R1}-C_{R2} = 2.72 \text{ \AA}$$

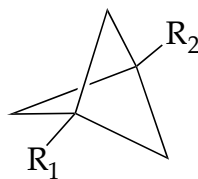


$$C_{R1}-C_{R2} = 2.79 \text{ \AA}$$

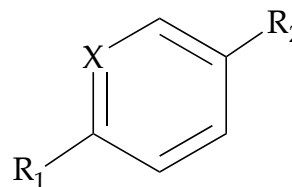
Other phenyl bioisosteres



$$C_{R1}-C_{R2} = 2.2 \text{ \AA}$$



$$C_{R1}-C_{R2} = 1.7 \text{ \AA}$$

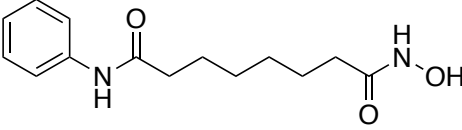
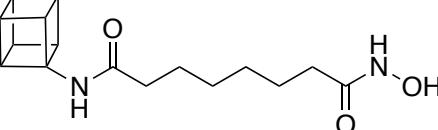
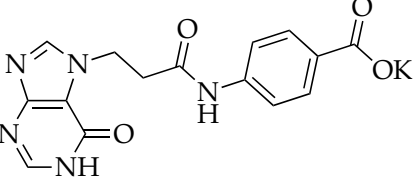
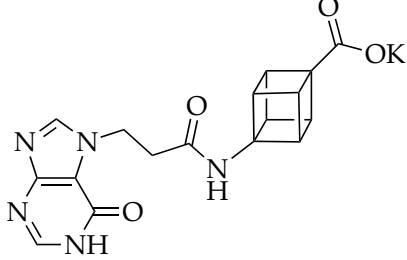
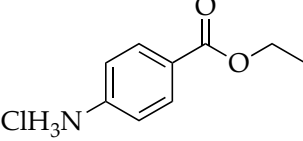
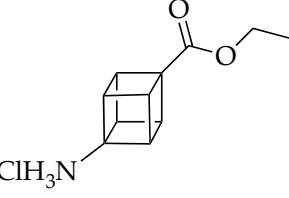
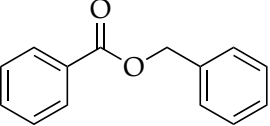
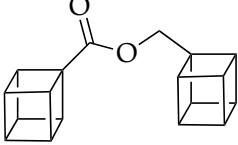
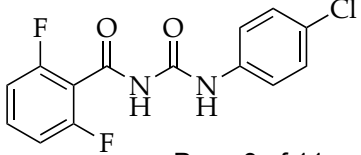
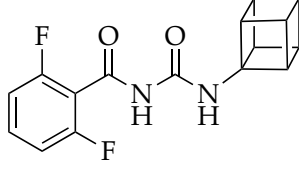


X=heteroatom substitution

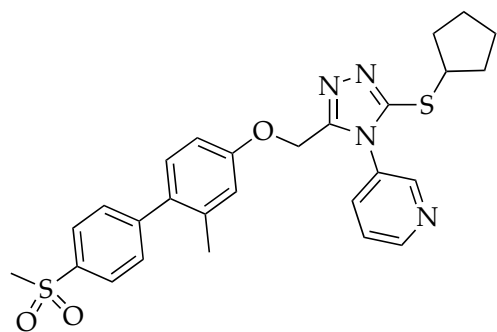
$$C_{R1}-C_{R2} = \text{varies on heterocycle}$$

1. Wager, T. T.; *et al. J Med Chem* **2011**, *54* (21), 7602-20.
 2. Stepan, A. F.; *et al. J Med Chem* **2012**, *55* (7), 3414-24.
- Mike Houghton @ Wipf Group

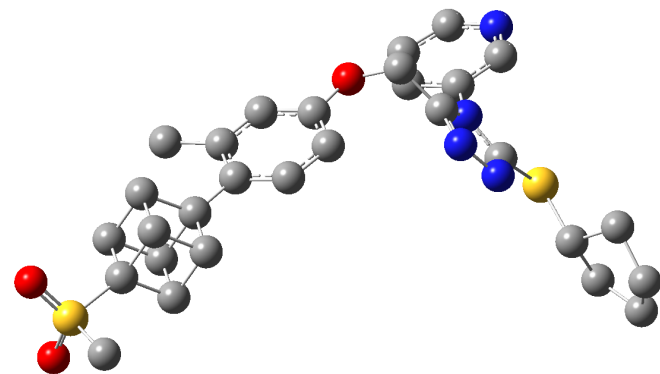
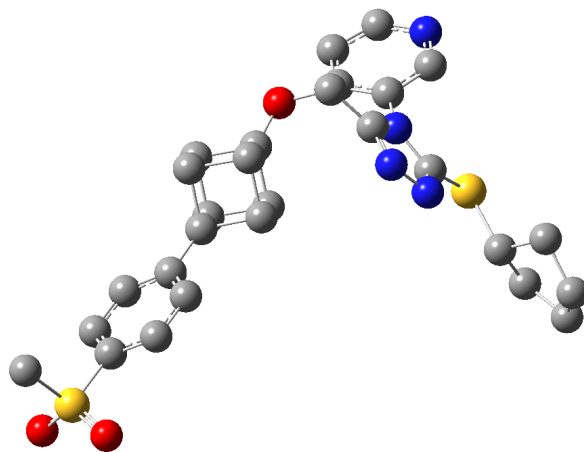
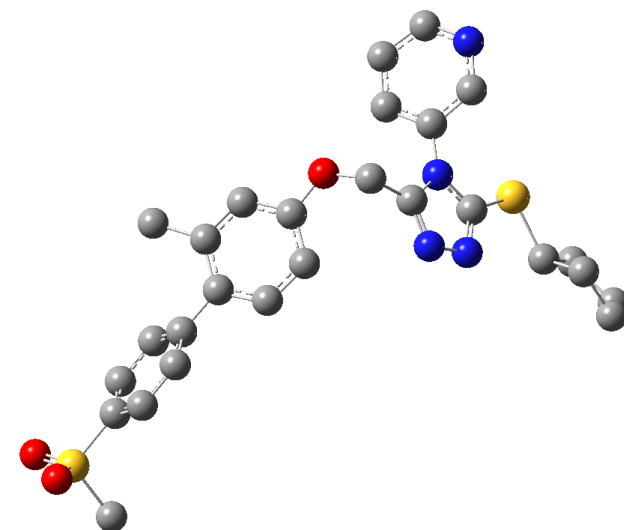
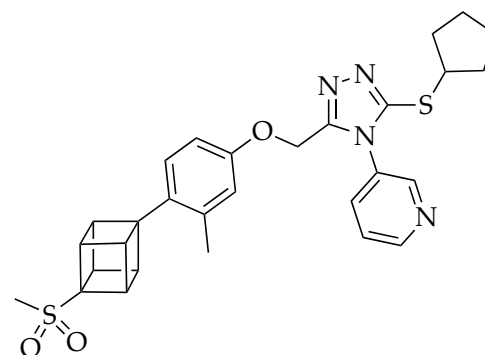
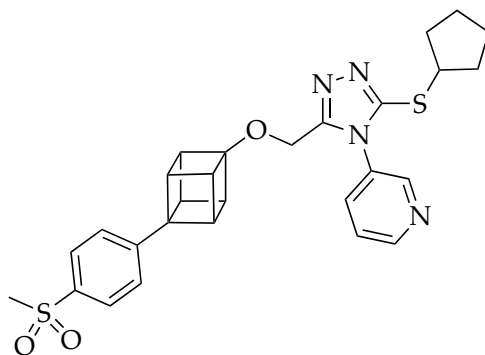
Highlight of Work Presented

Activity	pharmaceutical or agrochemical compound	corresponding cubane derivative
Same (IC ₅₀)	 <p>suberanilohydroxamic acid (SAHA)</p>	
Increase (neuronal differentiation capacity)	 <p>Leteprinin</p>	
Same (paw thermal threshold)	 <p>Benzocaine</p>	
Decrease (efficacy)	 <p>Benzyl benzoate</p>	
Increase (efficacy)	 <p>Diflubenzuron</p>	

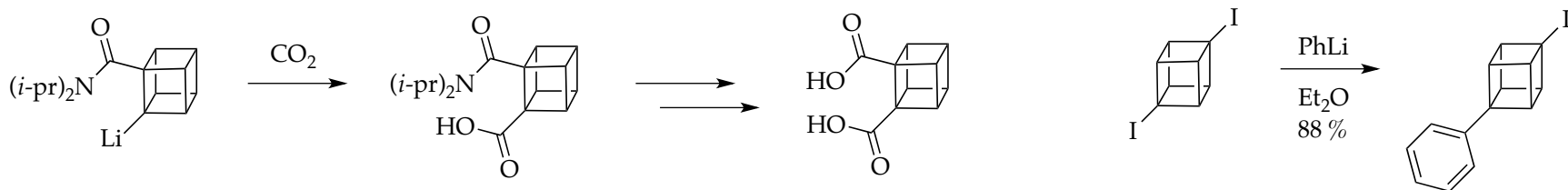
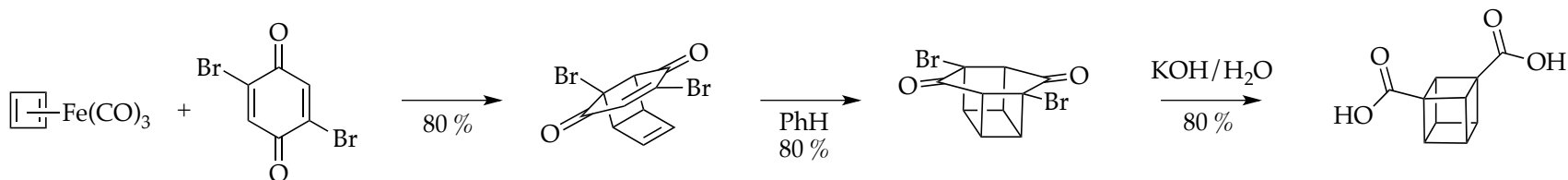
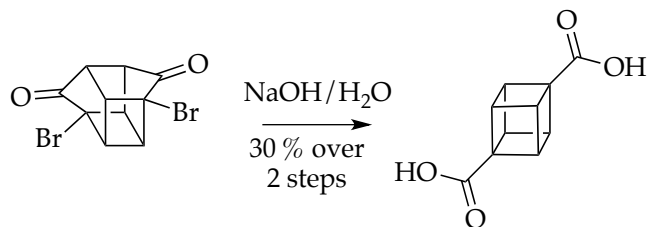
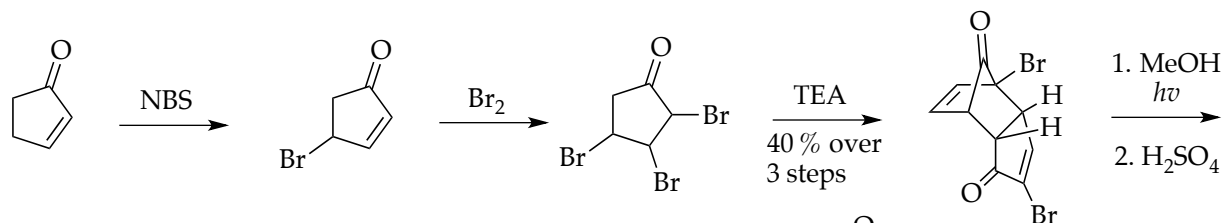
DFT Computations are Compelling



NMS-873



Cubane Synthesis



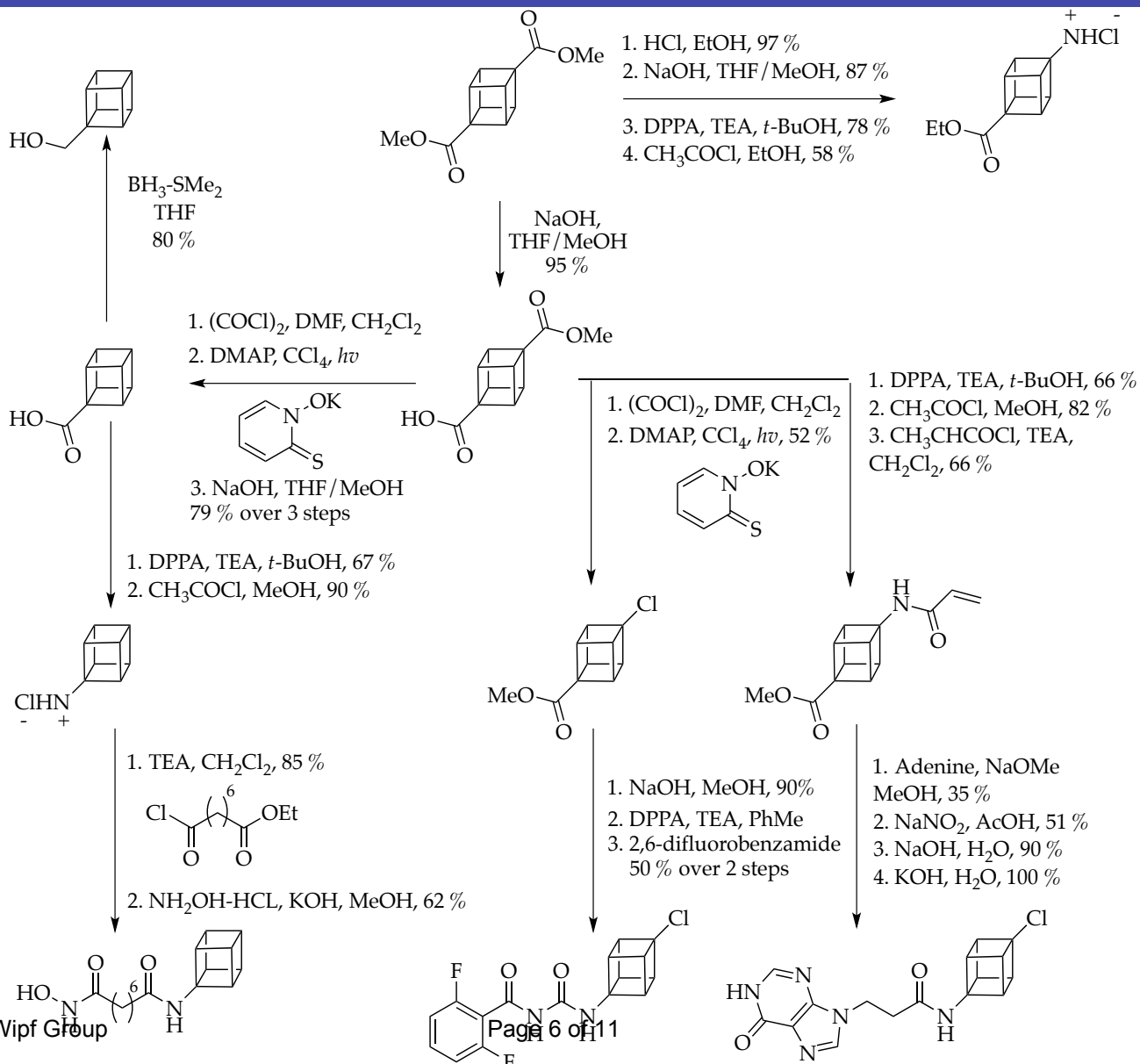
1. P. E. Eaton, T. W. Cole, Jr., *J. Am. Chem. Soc.* **1964**, *86*, 3157 – 3158.

2. P. E. Eaton, *Angew. Chem. Int. Ed. Engl.* **1992**, *31*, 1447–1462.

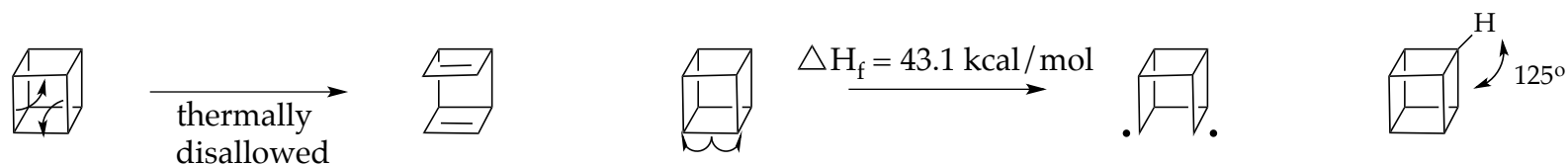
3. Statton, G. L.; Ramey, K. C. *J. Am. Chem. Soc.* **1966**, *86*, 1328 – 1329.

4. Wochal, J.; Davies, R. D.; Burton, J. *Org. Lett.* **2014**, *16* (16), 4094–7.

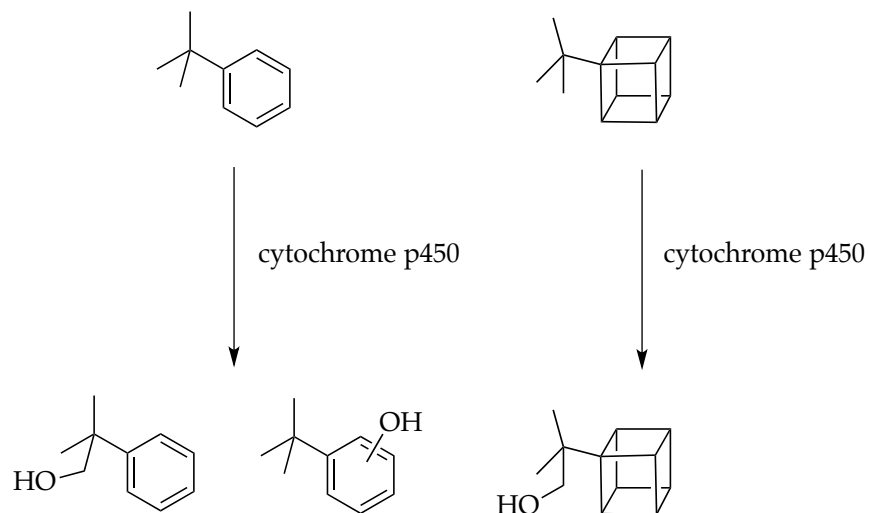
Cubane Transformations in This Paper



High Stability and Metabolic Stability of Cubane

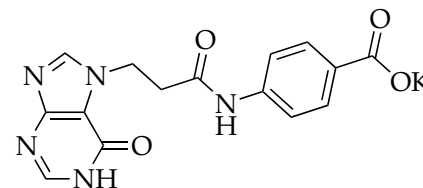
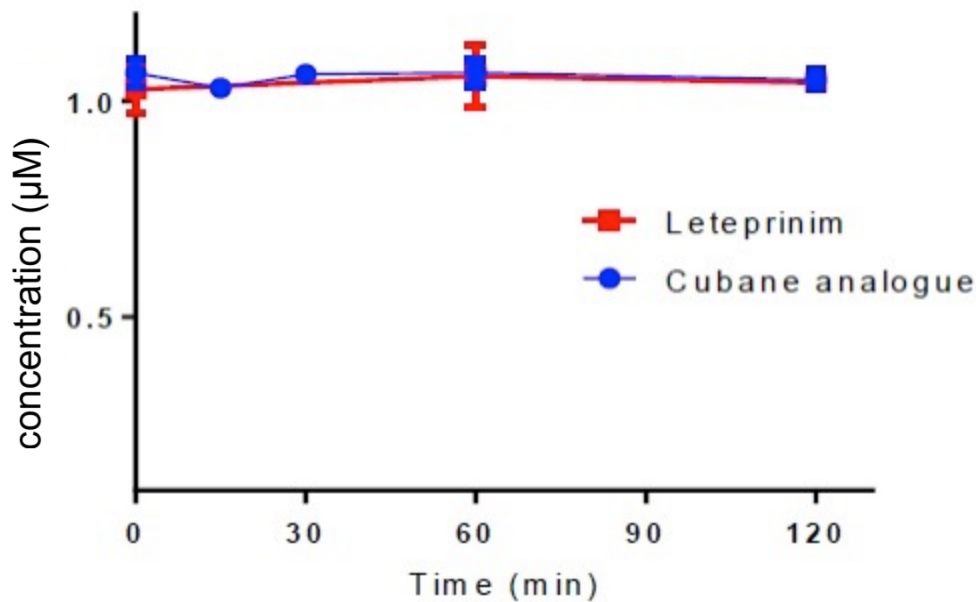


cytochrome p450

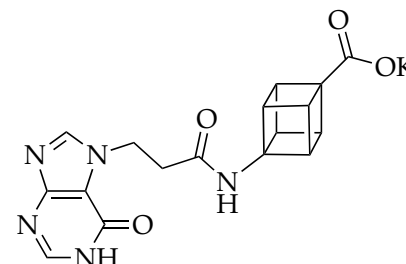


High Stability and Metabolic Stability of Cubane

In vitro human liver microsomes showed no metabolism of either the phenyl or cubane derivative

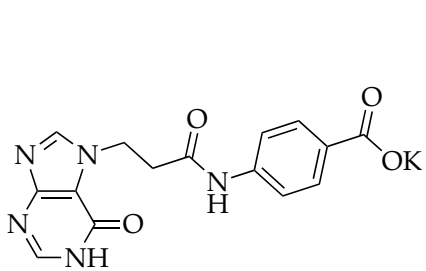


Leteprinim

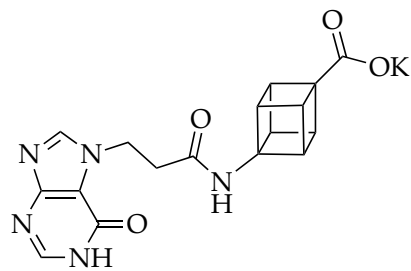


Letepricube

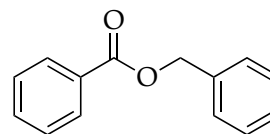
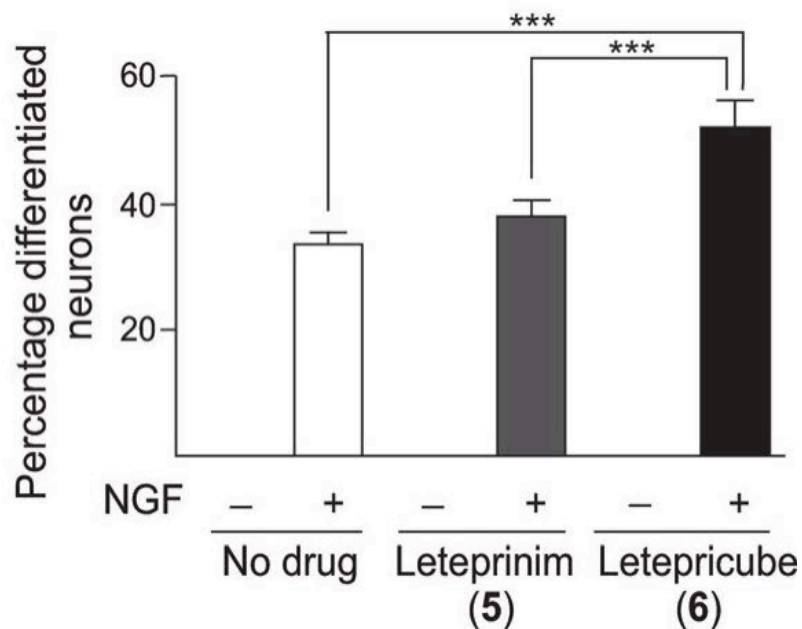
Advantages and Disadvantages of Cubane



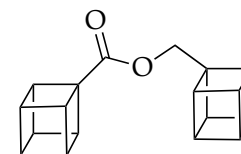
Leteprinim



Letepricube



Benzyl benzoate
LogP = 3.86



Cubyl cubates
LogP = 5.43

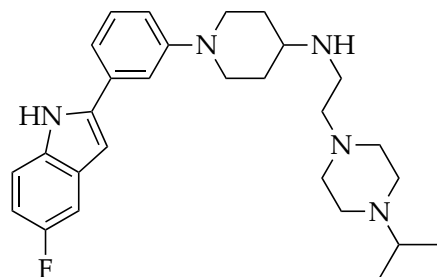
- The decrease in efficacy with the cubane derivative is traced to a lower solubility. This result suggests solubility matching must be taken into account.

-The increased activity of Letepricube could be attributed to its increased lipophilicity complimenting the CNS target's hydrophobic environment.

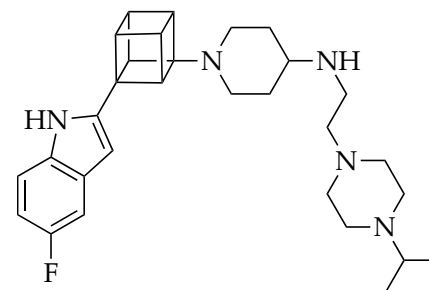
Conclusions

- Successfully showed cubane is a competent bioisostere for a phenyl group, thus validating Eaton's hypothesis
- Synthesized cubane derivatives of molecules with a wide range of applications
- Identified practical considerations when designing this isostere replacement
- Although convincing, the examples in the paper focused exclusively on *para* substituents

Possible Future Directions



UPCDC30245 °
 $C_{Ph1}-C_{Ph3} = 2.45 \text{ \AA}$



$C_{C1}-C_{C3} = 2.22 \text{ \AA}$ °

